ANIONIC OXYCOPE REARRANGEMENTS IN ACETYLENIC SPIRO SYSTEMS 1

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Abstract: The anionic oxy-Cope rearrangement of the acetylenic carbinol 3 furnishes the tricyclic diketone 5; substitution at the key position furnishes only the bicyclic 1,3-bridged dione 6.

In oxy-Cope rearrangements of neutral substrates, the substitution of a C=C for one of the olefinic moieties has been shown to accelerate such rearrangements. It was of interest therefore to study the effect of such a substitution in anionic oxy-Cope rearrangements. We report herein two anionic rearrangements of an acetylenic oxy-Cope moiety situated in a spiro system.

The spirodiones $\underline{1}$ and $\underline{2}$ were synthesized as per the reported procedures^{4,5}. Addition of lithium acetylide in liquid ammonia at -78°C for 2 h to the spiro dione $\underline{1}$ followed by work up gave the ethynyl carbinol $\underline{3}$ as a colourless crystalline solid⁶ (m.p. 124-125°C).

Treatment of the ethynyl carbinol 3 with potassium hydride in 1,2-dimethoxyethane at room temperature for 2 h afforded, after workup and purification, a colourless crystalline solid (m.p. 104-105°C) in 55% yield. Spectral data coupled with elemental analysis favour structure 5 for the rearranged product. The formation of compound 5 can be rationalised as in scheme I.

Scheme 1

In the above mechanism a flip of the cyclohexane ring to a more favourable diequatorial geometry is envisaged for the concerted reaction to occur followed by an isomerisation and a transannular Michael addition. It was of interest to study the rearrangement of compound 4 which contains a methyl group at a key atom (*) involved in the isomerisation step in the above sequence.

Addition of lithium acetylide in liquid ammonia at -78°C for 2 h to the spirodione $\underline{2}$ followed by workup gave the ethynyl carbinol $\underline{4}$ as a colourless crystalline solid (m.p. 102-103°C)⁷ in 45% yield.

Treatment of the ethynyl carbinol $\underline{4}$ with potassium hydride in 1,2-dimethoxyethane at 0°C for 1 h afforded after workup and purification a viscous liquid in 67% yield. Spectral data coupled with elemental analysis favour structure $\underline{6}$ for the rearranged product. Compound $\underline{6}$ is probably formed by a mechanism similar to the one pictured in Scheme 1; since isomerisation in species $\underline{7}$ (R=Me) is not possible the transannular Michael addition does not take place and the product is the product of protonation of species $\underline{7}$ (R = Me) itself viz the 1,3-bridged dione $\underline{6}$.

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- All the new compounds had the expected analytical values and IR and PMR absorptions.

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